

# Validation of radiochemical methods for the determination of difficult-to-measure nuclides using LSC

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$^{90}\text{Sr}$

$^{93}\text{Zr}$

Major issues in validation of  $^{63}\text{Ni}$  and  $^{59}\text{Ni}$

# Difficult-to-measure (DTM) nuclides

## Properties:

long half-life - long-term effect, small specific activity

low amounts/activities

$\alpha$  particles, electrons ( $\beta$ , Auger), X-rays – absorption, self-absorption

Nuclide	Half-life		Decay mode	$\beta_{\text{Max}}$ keV	Specific activity Bq/g	Occurance/ field of application		
<sup>90</sup> Sr	29,1	y	$\beta^-$	546	5,04E+12	DN	WD	EM, MT
<sup>93</sup> Zr	1,53	10 <sup>6</sup> y	$\beta^-$	90	9,27E+07	DN	WD	
<sup>59</sup> Ni	76400	y	EC, $\beta^+$	(X: 6,9)	2,94E+09		WD	
<sup>63</sup> Ni	100	y	$\beta^-$	67	2,09E+12	DN	WD	

DN=decommissioning of nuclear facilities

WD=waste depository

EM=environmental monitoring

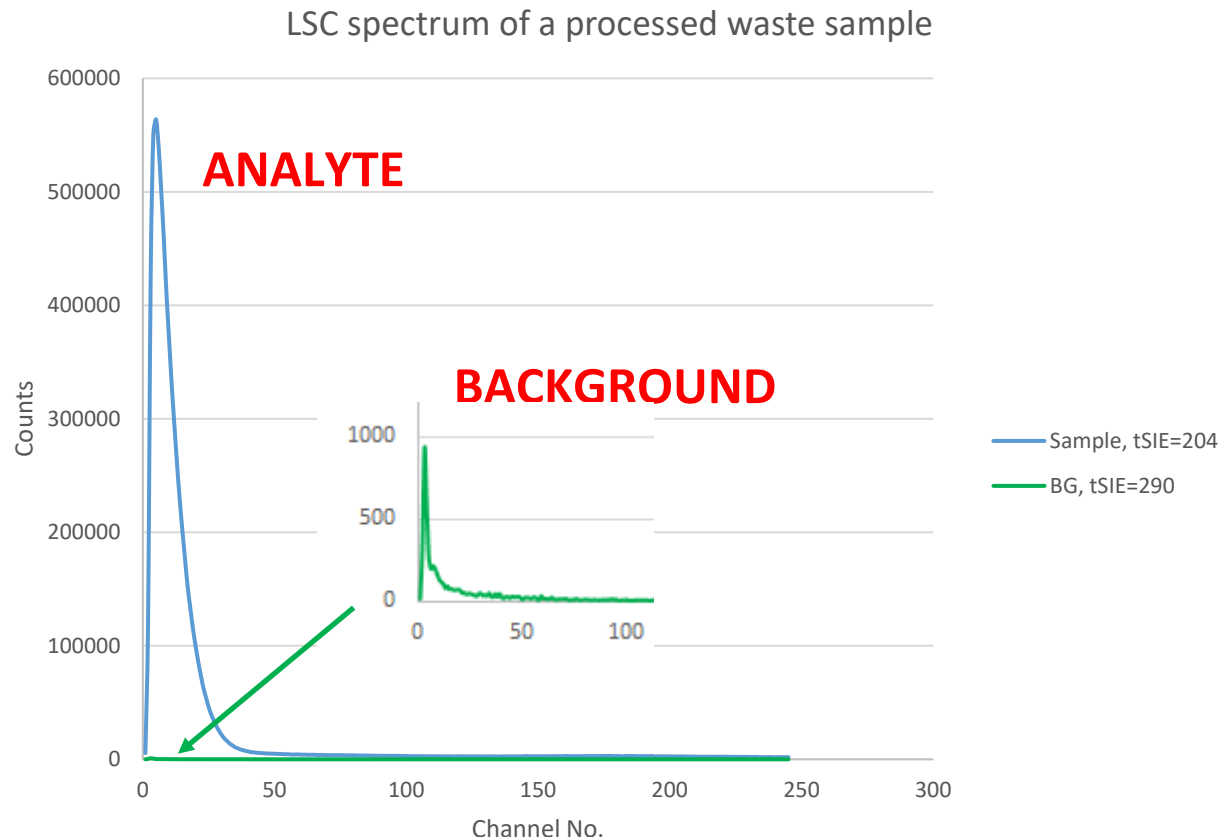
MT=medical tracer

# $\beta$ spectrometry by LSC

High detection efficiency (close to 100 %)  
No absorption/self-absorption



Quench effect – quench correction possibilities  
Poor energy resolution (around 10 %)  
**Continuous nature of  $\beta$  decay/spectra**

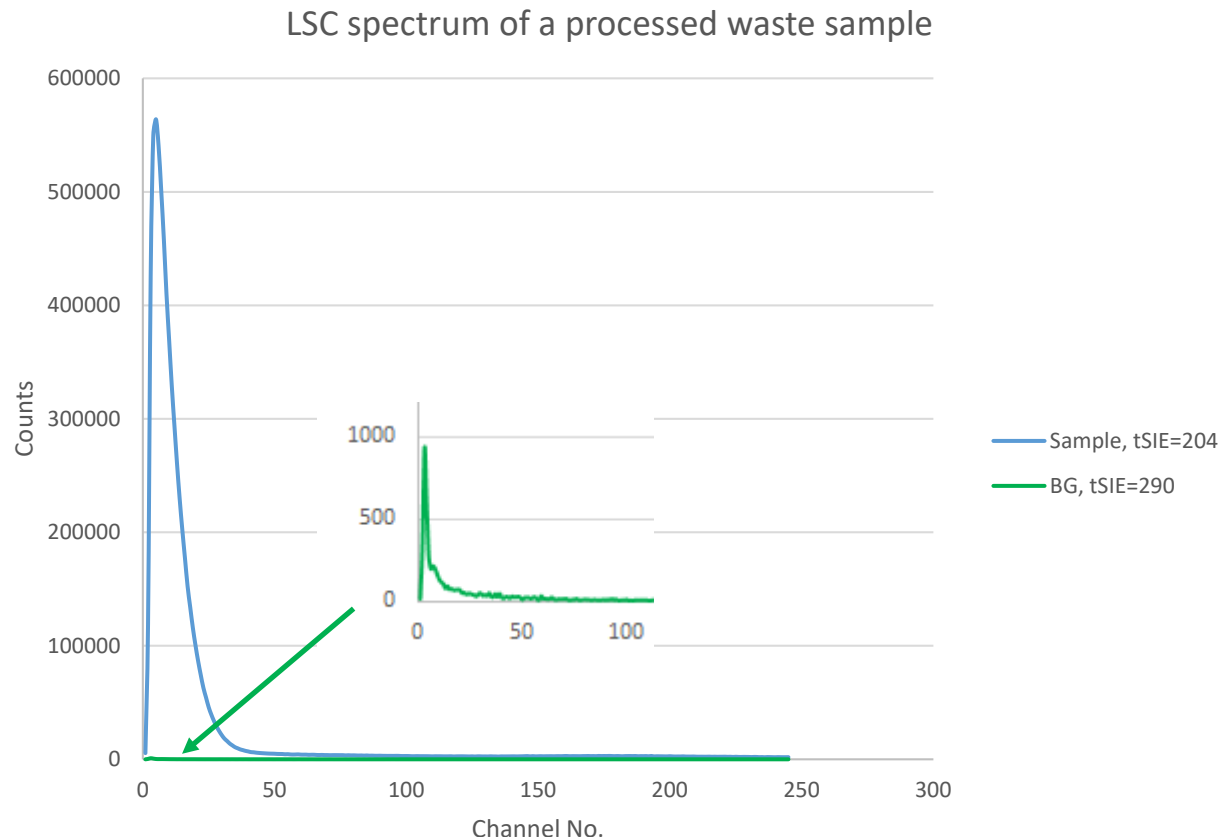


Which analyte?  
 $^3\text{H}$  – 18 keV  
 $^{63}\text{Ni}$  – 67 keV  
 $^{151}\text{Sm}$  – 77 keV  
 $^{93}\text{Zr}$  – 90 keV  
 $^{85}\text{Sr}$ ,  $^{59}\text{Ni}$ ,  $^{55}\text{Fe}$  ...

# LSC measurement

Chemical separation procedure

High selectivity - to remove interfering nuclides



Which analyte?

$^3\text{H}$  – 18 keV

**$^{63}\text{Ni}$  – 67 keV**

$^{151}\text{Sm}$  – 77 keV

$^{93}\text{Zr}$  – 90 keV

$^{85}\text{Sr}$ ,  $^{59}\text{Ni}$ ,  $^{55}\text{Fe}$  ...

# Method validation

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by either or the combination of any the following techniques

1. calibration using reference standards or reference materials,
2. inter-laboratory comparisons, e.g. PT
3. systematic assessment of the factors influencing the result,
4. assessment (of the uncertainty of the results) based on scientific understanding of the theoretical principles of the method and practical experience,
5. comparison of results achieved with other methods

ISO/IEC 17025 (2005): General requirements for the competence of testing and calibration laboratories

# $^{90}\text{Sr}$ determination

Nuclide	Half-life		Decay mode	$\beta_{\text{Max}}$ keV	Specific activity Bq/g
$^{90}\text{Sr}$	29.1	y	$\beta^-$	546	5,04E+12
$^{90}\text{Y}$	64	h	$\beta^-$	2280	2,01E+16

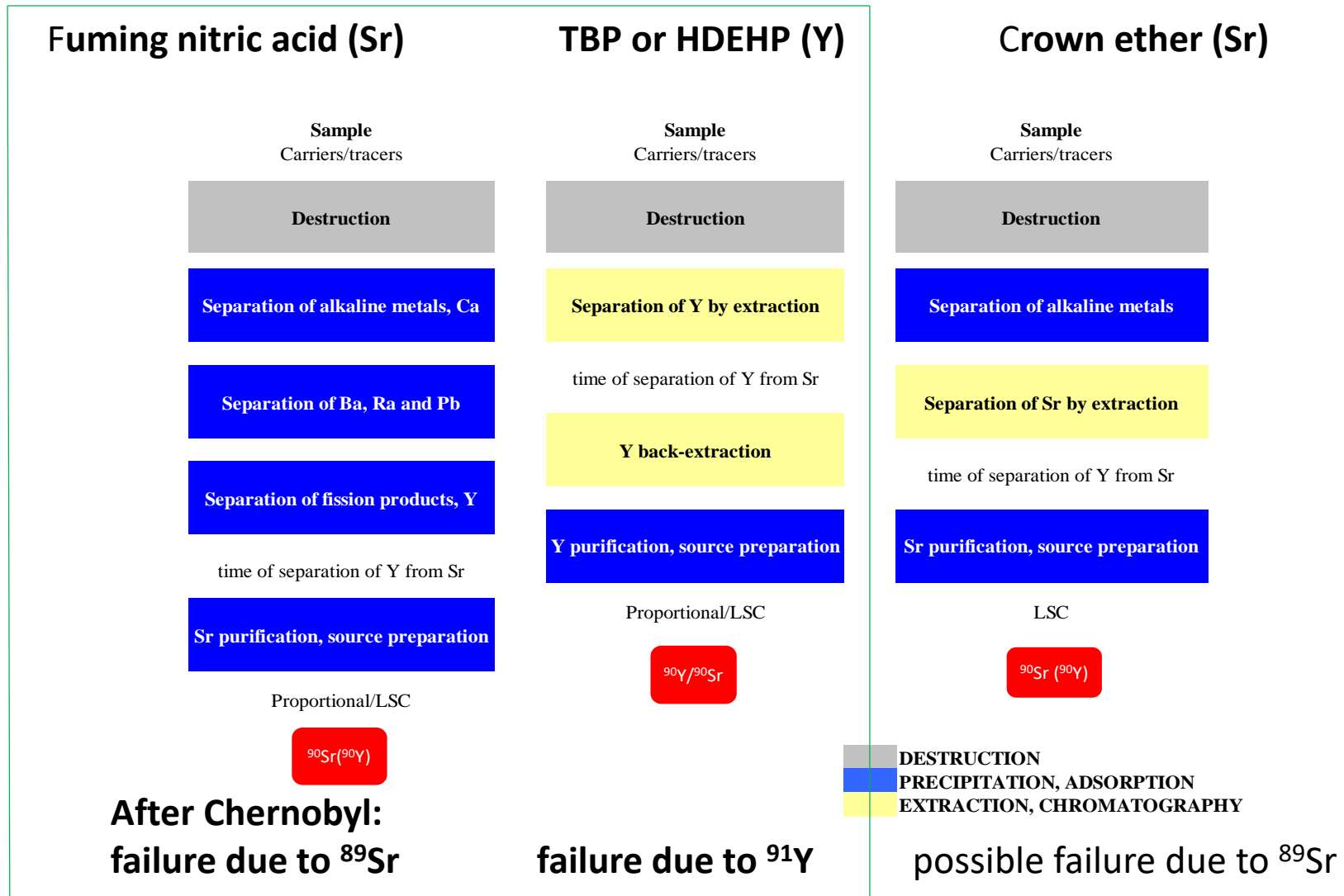
## Standard validation techniques:

1. Use of **CRM**,
2. Participation in **PTs**
3. Systematic assessment of the factors influencing the result - fish bone
4. Assessment based on scientific understanding

## Validation issue

5. No good independent method for validation

# $^{90}\text{Sr}$ determination

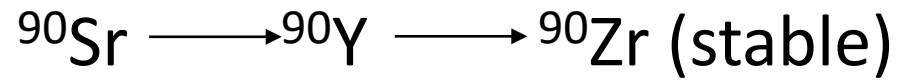


Methods validated under normal conditions can fail during accidents!

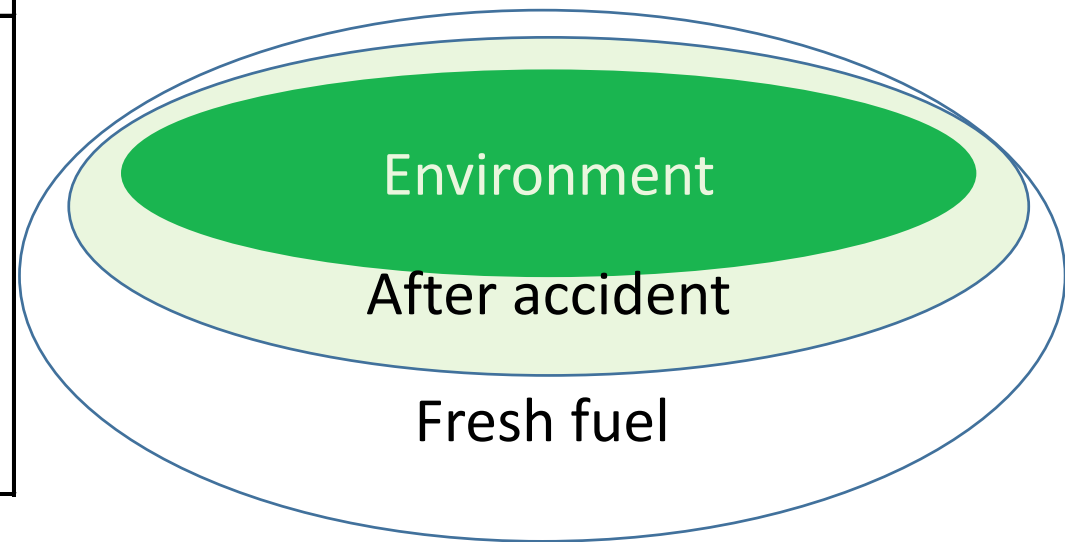


# $^{90}\text{Sr} / ^{89}\text{Sr}$ determination

Analytes and possible interferences in radiostrontium analysis:

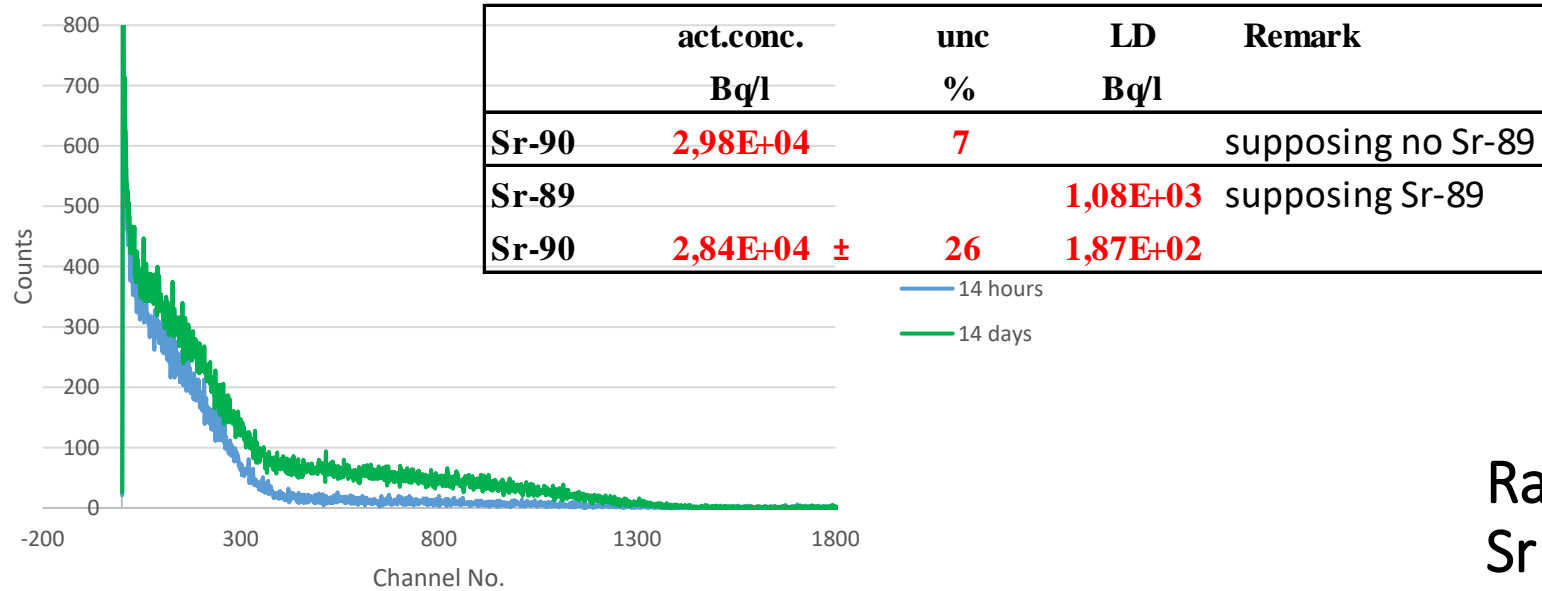


Nuclide	Half-life		Decay mode	$\beta_{\text{Max}}$ keV	Specific activity Bq/g
$^{89}\text{Sr}$	50.6	d	$\beta^-$	1501	5,04E+12
$^{90}\text{Sr}$	29.1	y	$\beta^-$	546	
$^{90}\text{Y}$	64	h	$\beta^-$	2280	
$^{91}\text{Y}$	59	d	$\beta^-$	1544	
$^{91}\text{Sr}$	9,7	h	$\beta^-, \gamma$	2700	
$^{92}\text{Sr}$	2,6	h	$\beta^-, \gamma$	568	



## Old waste

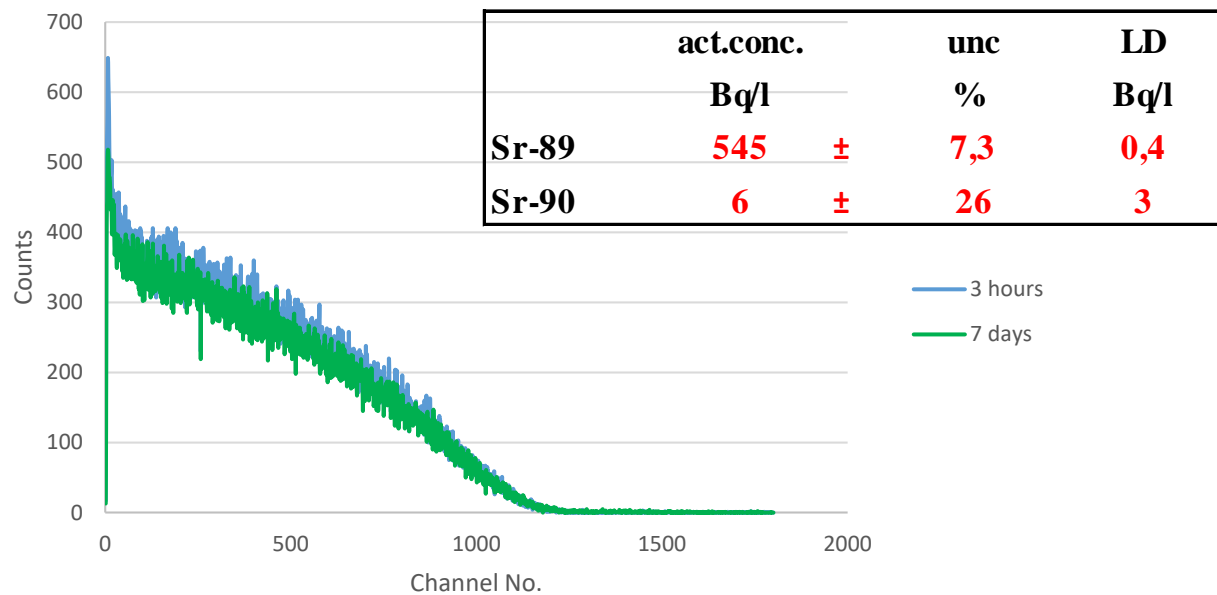
<sup>90</sup>Sr LSC spectrum of H16-3 waste sample



## Radiostrontium determination using Sr crown ether in nuclear samples

<sup>89</sup>Sr, <sup>90</sup>Sr LSC spectrum of 10TV20 primary coolant sample

## Fresh coolant



- separation after 2-5 days of **cooling**
- **repeated counting** after 2 weeks
- checking the shape of LS spectra
- calculations described in IAEA-AQ-27\_web

# Validation of the method for Radiostrontium determination

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## Conclusions

$^{90}\text{Sr}$  results can be validated by measurement of reference materials, by participation in inter-laboratory comparisons.

It is important to assess all influencing factors, such as the presence of other Sr and Y isotopes, or other possible contamination (radionuclides of K, Ba, Pb) by performing ingrowth/decay measurements as part of method validation.

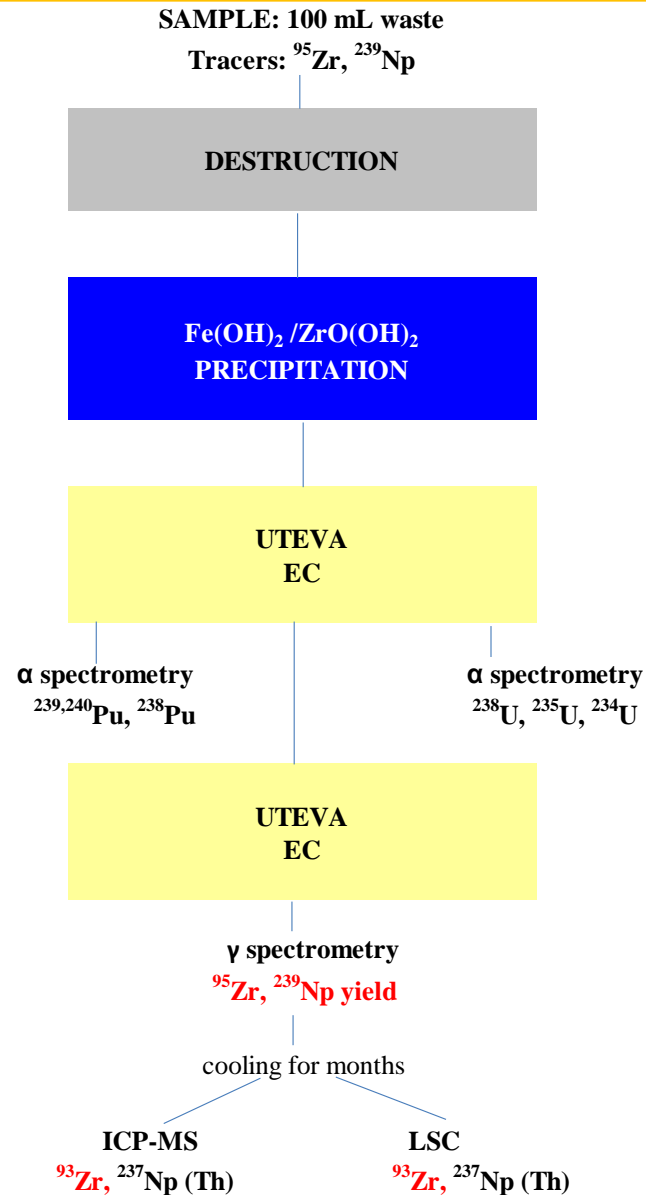
# $^{93}\text{Zr}$ determination

Nuclide	Half-life	Decay mode	$\beta_{\text{Max}}$ keV	Specific activity Bq/g	Occurance/ field of application
$^{93}\text{Zr}$	1,53 10 <sup>6</sup> y	$\beta^-$	90	9,27E+07	DN WD

**Validation issues:** No  $^{93}\text{Zr}$  standard  
No reference material  
No intercomparison exercise

**Validation option:**  
comparison of independent methods: **LSC**  
**ICP-MS**

# $^{93}\text{Zr}$ determination



**Carrier free tracers by NAA of U**  
Purification on UTEVA  
 $\gamma$  spectrometry

**Repeated separation on UTEVA**  
to purify Zr

**Yield:** Zr: 60-90%  
Fluctuating significantly due to Zr  
adsorption to surfaces.

# $^{93}\text{Zr}$ determination

## **ICP-MS** (Agilent Triplequad 8800)

Possible interferences in ICP-MS

isobaric interference: **stable Nb** ( $^{93}\text{Nb}$ ),  $^{93\text{m}}\text{Nb}$ ,  $^{93}\text{Mo}$

abundance sensitivity: **stable Zr** ( $^{92}\text{Zr}$ ,  $^{94}\text{Zr}$ )

polyatomic interferences, e.g.  $^{92}\text{Zr}^1\text{H}$

## **LSC** (Perkin Elmer Tricarb 2800)

Possible interferences in LSC

**$^{95}\text{Zr}$ - $^{95}\text{Nb}$  tracer**

$\alpha$  emitting nuclides ( $^{237}\text{Np}$ ,  $^{232}\text{Th}$ ,  $^{230}\text{Th}$ ...)

**any contamination** ( $^{110\text{m}}\text{Ag}$ ...  $^{93\text{m}}\text{Nb}$ )

## **Countermeasures**

Nb decontamination  $\text{DF} > 10^3$

mathematical correction

mathematical correction

cooling, correction

spectrometric separation

chemical purification

# Measurement of $^{93}\text{Zr}$ by ICP-MS

- Background correction
- Correction for measurement instability using internal standard (Rh)

$$I_{net} = (I_{measured} - I_{blank}) \cdot \frac{I_{Rh,ref}}{I_{Rh}}$$

- Correction for abundance sensitivity due to stable Zr ( $^{92}\text{Zr}$ ,  $^{94}\text{Zr}$ )

$$I_{net,cor.} = I_{net} - I_{net,Zr92+94} \cdot f_1 \quad f_{(93/92+94)}: 4.4 \cdot 10^{-5} \text{ cps/cps}$$

- Calibration with stable Zr standard solution ( $^{92}\text{Zr}$ )

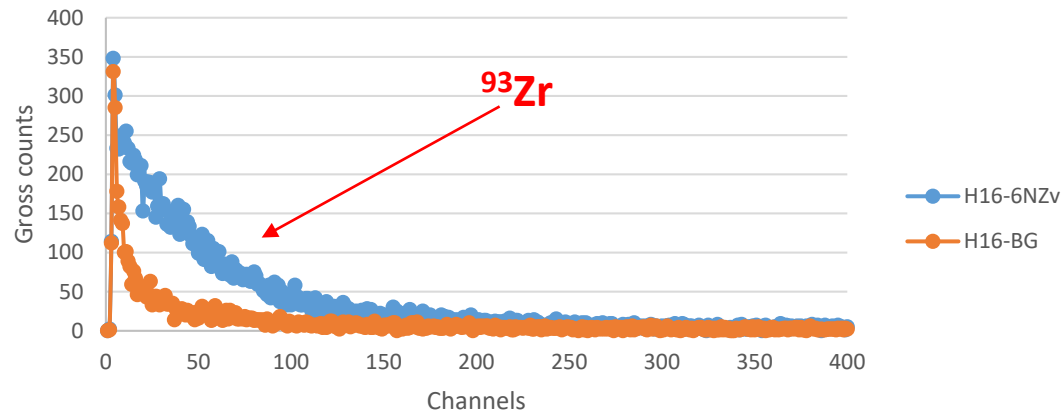
$$C_{Zr-93} = \frac{I_{net,cor}}{I_{net,cor,Zr-92std}} C_{Zr-92std}$$

sensitivity:  $5.6 \cdot 10^4$  cps/ppb nuclide

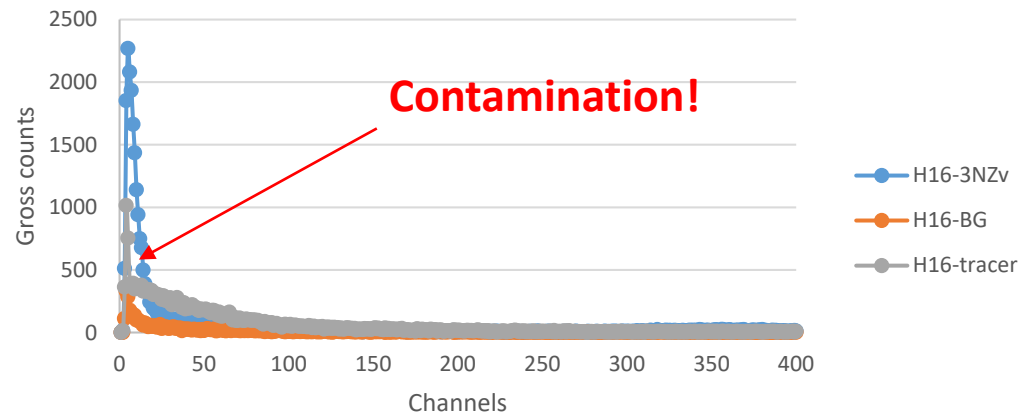
- No correction for isobaric Nb interference

# Measurement of $^{93}\text{Zr}$ by LSC

LSC spectra of H16-6 waste sample and the background



LSC spectra of H16-6 waste sample, the  $^{95}\text{Zr}$  tracer and the background

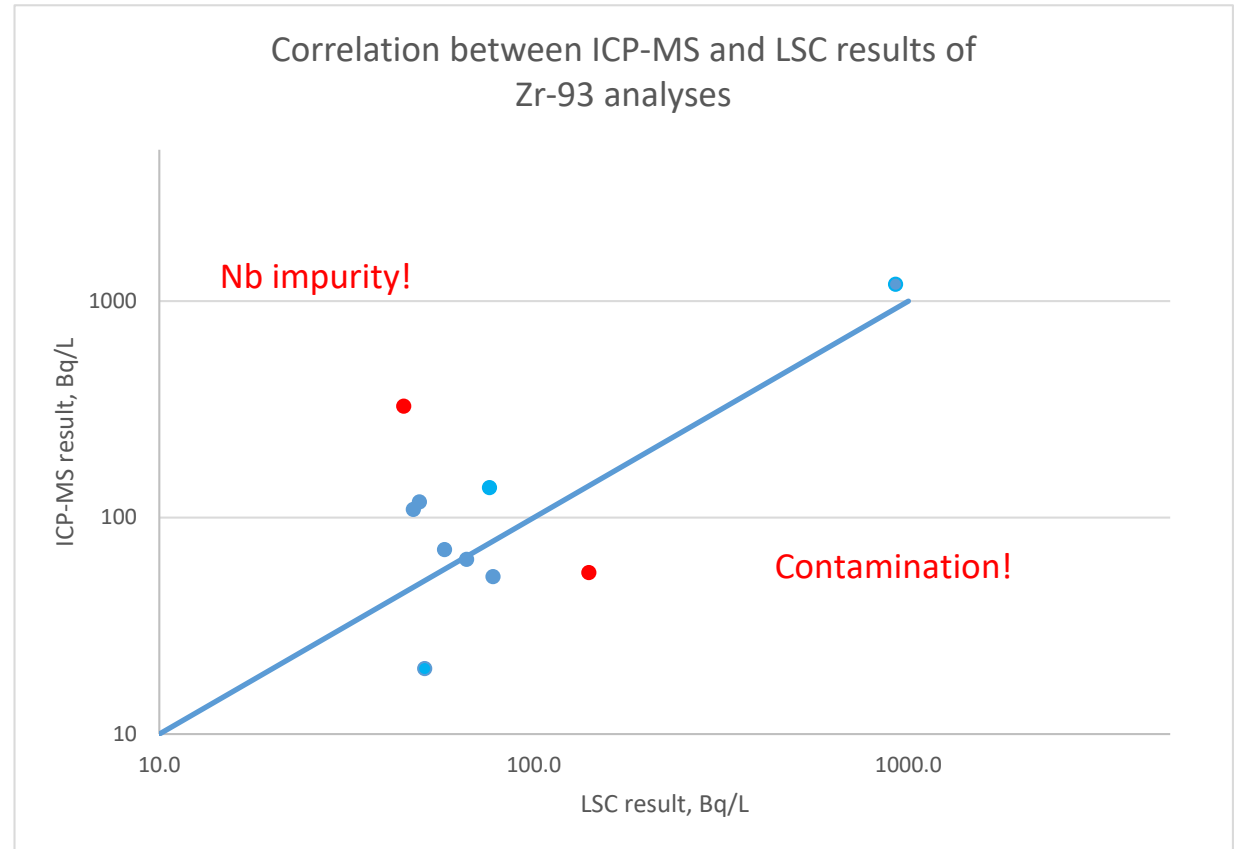


LS spectra can reveal radioactive contamination of Zr fraction!



# Measurement of $^{93}\text{Zr}$ by ICP-MS and LSC

Sample code	Zr-93		Remark
	LSC	ICPMS	
	Bq/L	Bq/L	
H16-10	51,1	20,0	
H16-1	77,9	53,3	
H16-9	57,8	71,0	
H16-8	47,8	108,8	
H16-5	49,6	118,0	
H16-2	76,1	137,4	
H16-4	44,9	327,0	Nb impurity
H16-6	924,8	1196,3	
H16-3	140,3	55,7	contamination



# Measurement of $^{93}\text{Zr}$ by ICP-MS and LSC

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## Conclusions

ICP-MS is a sensitive technique to determine  $^{93}\text{Zr}$  but small amounts of stable Nb can cause overestimation of the result.

LSC is exposed to the hazards of radioactive contamination. The presence of radioactive Zr tracer reduces the sensitivity.

LSC helps discover deviation from true value.

The method can be validated by the use of two methods.

# $^{63}\text{Ni}$ and $^{59}\text{Ni}$ determination

Nuclide	Half-life		Decay mode	$\beta_{\text{Max}}$ keV	Specific activity Bq/g	Occurance/ field of application		
$^{59}\text{Ni}$	76400	y	EC, $\beta^+$ (X: 6,9)		2,94E+09		WD	
$^{63}\text{Ni}$	100	y	$\beta^-$	67	2,09E+12	DN	WD	

Validation issues: No  $^{59}\text{Ni}$  standard

No reference material for  $^{59}\text{Ni}$  determination

No intercomparison exercise for  $^{59}\text{Ni}$  determination

Validation options:

- $^{63}\text{Ni}$  determination in CRM and in PT

- Comparison of independent methods:

**LSC for  $^{63}\text{Ni}$ ,**

**X spectrometry for  $^{59}\text{Ni}$**

**ICP-MS for  $^{63}\text{Ni}$  ???,  $^{59}\text{Ni}$ ???**

# $^{63}\text{Ni}$ and $^{59}\text{Ni}$ determination

**SAMPLE**  
Carriers: 6 mg Fe, 6 mg Ni

**DESTRUCTION**

**MIBK**  
**EC**

LSC/X spectrometry  
 $^{55}\text{Fe}$

Removal of Fe by extraction chromatography

**DMG**  
**EC**

Selective separation of Ni by extraction chromatography (from Co, Cu)

**DMG precipitate**

Ni purification by precipitation

X spectrometry

$^{59}\text{Ni}$

LSC

$^{63}\text{Ni}$

Decomposition of DMG

(Fe, Ni recovery: AAS)

# $^{63}\text{Ni}$ and $^{59}\text{Ni}$ determination

## LSC for $^{63}\text{Ni}$ (Perkin Elmer Tricarb 2800)

Possible interferences

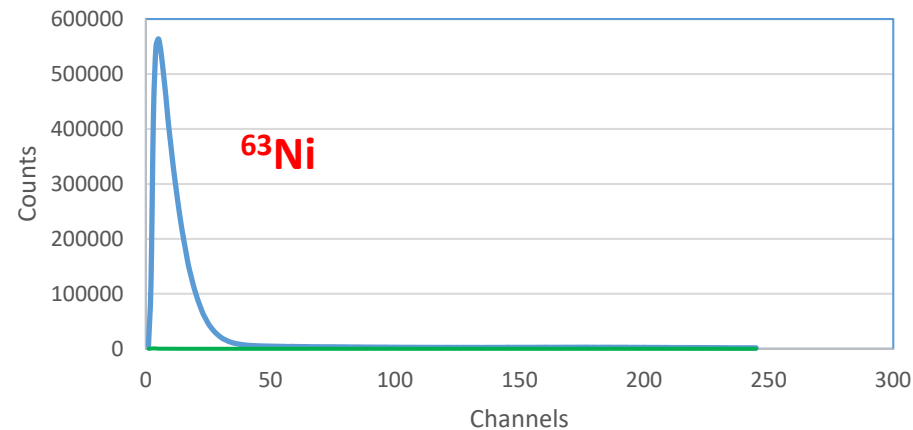
$^{59}\text{Ni}$

correction if necessary

any contamination ( $^{110\text{m}}\text{Ag}$ ...  $^{60}\text{Co}$ )

chemical purification

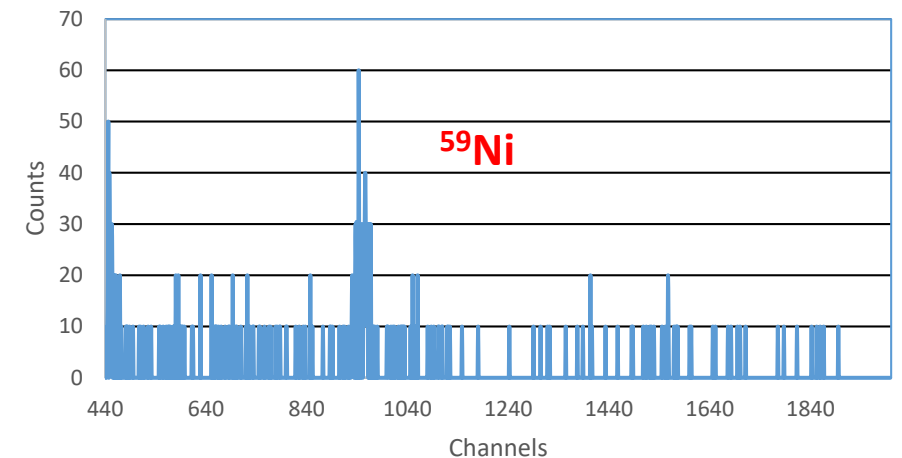
LSC spectrum of a Ni source from H16-3 waste



## X-ray spectrometry for $^{59}\text{Ni}$ (Si(Li) and MCA)

Possible interferences: none

X-ray spectrum of a Ni source from H17-4 waste



# $^{63}\text{Ni}$ and $^{59}\text{Ni}$ determination

**ICP-MS** (Agilent Triplequad 8800)

Possible interferences in ICP-MS

isobaric interference of  $^{63}\text{Ni}$ : **stable Cu** ( $^{63}\text{Cu}$ - $^{65}\text{Cu}$ )

isobaric interference of  $^{59}\text{Ni}$ : **stable Co** ( $^{59}\text{Co}$ )

abundance sensitivity: stable Ni ( $^{58}\text{Ni}$ ,  $^{60}\text{Ni}$ ,  $^{62}\text{Ni}$ )

polyatomic interferences, e.g.  $^{58}\text{Ni}^1\text{H}$ ,  $^{62}\text{Ni}^1\text{H}$

Calculation: similar to the case of  $^{93}\text{Zr}$

## Countermeasures

mathematical correction

chemical purification

mathematical correction

collision cell

# Validation of <sup>63</sup>Ni results

## Use of reference materials, participation in PTs

	Measured		Reference value		Precision
	<sup>63</sup> Ni activity concentration	unc	<sup>63</sup> Ni activity concentration	unc	index
	Bq/kg	Bq/kg	Bq/kg	Bq/kg	%
MAPEP-06-MaS15	357	50	329	56	22

Not waste matrix

## Measurement by ICP-MS

Sample code	ICP-MS						LSC	ppb	
	Net intensity at mass 63 (Ni-63+Cu-65)	Net intensity at mass 65 (Cu-65)	Ni-63 intensity corrected for Cu-63	Ni-63 concentration	Ni-63 activity concentration	Ni-63 activity concentration	Ni-63 activity concentration	Cu concentration in waste*	Ni concentration in waste
	cps	cps	cps	ppb	Bq/mL	Bq/L waste	Bq/L waste	ppb	ppm
H16-3/1	3,14E+07	1,48E+07	LD	LD	LD	LD	1,38E+05	1054	11
H16-3/2	5,15E+07	2,43E+07	LD	LD	LD	LD	1,26E+05	1328	9
H17-4	7,69E+05	3,64E+05	LD	LD	LD	LD	2,99E+05	107	28
H17-6	1,92E+06	8,28E+05	8,55E+04	1,53E+00	3,19E+03	2,87E+06	6,42E+05	76	37

The presence of a few ppb Cu causes significant overestimation!

Cu-63/Cu-65 ratio: 2,22

Ni sensitivity: 5,61E+04 cps/ppb nuclide

Ni-63 specific activity: 2,09E+03 Bq/ng

\* in ICP-MS sam

Work in high purity environment

# Validation of <sup>59</sup>Ni results

No reference material, no PT

Measurement by ICP-MS

Sample code	MP-AES measurement	ICP-MS measurements					XRF measurements
	Co concentration by AES ppb	Net intensity at mass 59 cps	Ni-59 intensity corrected for Co cps	Ni-59 concentration ppb=ng/mL	Ni-59 activity concentration Bq/mL	Ni-59 activity concentration Bq/L waste	Ni-59 activity concentration Bq/L waste
H16-3/1	1,25E+02	5,10E+06	LD	LD	LD	<b>LD</b>	<b>LD</b>
H16-3/2	3,10E-01	9,16E+04	7,47E+04	1,33E+00	3,92E+00	<b>2488</b>	<b>LD</b>
H17-4	LD	7,77E+04	7,77E+04	1,39E+00	4,08E+00	<b>4587</b>	<b>403</b>
H17-6	LD	1,21E+05	1,21E+05	2,15E+00	6,33E+00	<b>5694</b>	<b>1080</b>

Co sensitivity: 5,44E+04 cps/ppb

Ni sensitivity (isotopic): 5,61E+04 cps/ppb nuclide

Ni-59 specific activity: 2,94 Bq/ng

The presence of a 1-2 ppb Co causes significant overestimation!



# $^{63}\text{Ni}$ and $^{59}\text{Ni}$ determination

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## Conclusions

$^{63}\text{Ni}$  results obtained by LSC can be validated by the measurement of reference materials, participation in intercomparison exercises.

Techniques for validation of  $^{59}\text{Ni}$  results obtained by X-ray spectrometry are not available.

# MESSAGES



Conventional techniques of method validation are not always sufficient or available.

The accuracy of  $^{90}\text{Sr}$  results can be confirmed by performing repeated measurements.

$^{93}\text{Zr}$  results can be validated by comparing LSC and ICP-MS measurements.

Validation of  $^{63}\text{Ni}$  and  $^{59}\text{Ni}$  results by comparing LSC/X-ray spectrometry and ICP-MS measurements has failed due to the presence of a few ppb Cu or Co.